



CHROMIC ACID FLAKE CA ULTRA



Chromic Acid Ultra Flake is a premium quality product exceeding GSA Commercial Item Description specifications for Technical Grade chromic acid. It is specially processed to be virtually dust-free while still providing excellent storage, handling and quick dissolving qualities. The low sodium content is suitable for use in catalysts and other special applications.

Chemical Formula: CrO₃

Molecular Weight: 99.99

CAS Number: 1333-82-0

Appearance: Dark Reddish Purple Flakes

True Density: 2.7 @ 20 C (68 F)

Bulk Density: Loose: 68 - 75 lbs. per cubic foot

Stability in Air: Very hygroscopic and deliquesces at humidity above 30%

Solubility in Water: Highly soluble: 62.5% w/w at 20 C (69 F)

Heat of Solution: Exothermic (24.7 cal./g; 44.5 BTU/lb.)

Behavior on Heating: Melts at 197 C (387 F), starts to decompose to intermediate chromium oxides at slightly higher temperatures.

Chemical Characteristics: Chromic acid is a strong acid and oxidizing agent. When neutralized with alkalis, chromic acid forms dichromate or chromate compounds. In oxidation reactions the chromium atom is reduced from the hexavalent to trivalent state.

ANALYSIS:

	Specification	Typical Analysis*
CrO ₃	99.85 % Min.	99.9 %
Sulfate as SO ₄	0.10 % Max.	0.07 %
Chloride as Cl	0.005 % Max.	<0.005 %
Insoluble Matter	0.005 % Max.	0.002 %
Sodium as Na ₂ O	0.020 % Max.	0.014 %
pH of 1% Aqueous Solution	NA	1.0

*Data represent normal production values and may be slightly higher or lower for actual product lots.

GENERAL APPLICATIONS:

The major uses of chromic acid are in wood preservation, metal finishing and plating. Secondary applications include catalyst manufacture and as an oxidizing agent.

Chromic acid is used to produce salt-free CCA type wood preservatives. These preservatives are combinations of chromium, copper and arsenic. The CCA offers protection from wood destroying fungi, marine borers, termites, ascomycetes, white rot, dry rot and brown rot.

The two main types of chromium electroplating are decorative and functional. Decorative chromium plating is usually applied as a very thin layer to articles that have been nickel plated and is used for plumbing fixtures, automobile accessories, domestic appliances, etc., imparting a bright shiny finish. Functional or "hard" chromium plate is applied in much thicker layers and gives an extremely hard, corrosion-resistant surface with a low coefficient of friction. Hard chrome plating finds use in piston rings, engine cylinders, inspection and cutting tools and metal working machinery. The finish is usually dull. Electrodeposition of chromium from solutions containing chromic acid is also used in the manufacture of pure chromium metal for incorporation in creep resistant high temperature alloys.

Chromic acid is used in the anodizing of aluminum. The extremely thin layer of oxide, which gives aluminum metal its resistance to corrosion, is thickened by anodic oxidization using chromic acid solution as the electrolyte. The coatings obtained are semi-opaque and gray in color. They are capable of being dyed to give a very pleasing finish. They also provide a good base for the application of paint and other organic finishes.

Chromate or dichromate ions are widely used in the metal finishing field to impart corrosion resistance. Conversion coatings, chromate containing films of oxide, are formed on metal surfaces by using sodium dichromate, potassium dichromate or chromic acid. Chromic acid solutions are also used for sealing phosphate coatings applied to iron or steel to improve their corrosion resistance. Other applications in the metal finishing field include its use for bright dipping of brass and acid cleaning of aluminum and magnesium.

Chromic acid is a powerful oxidizing agent, which leads to a variety of uses in organic syntheses and in the bleaching of oils, fats and waxes. Chromic acid is used in the manufacture of other chromium compounds and as an analytical reagent.

HEALTH AND SAFETY PRECAUTIONS:

Chromic acid is a hexavalent chromium compound, and contact with it, either in the solid or solution form, should be avoided. In case of contact, flush thoroughly with water. For eyes, give prolonged irrigation with water, and get medical attention immediately.

For detailed information, consult our Safety Data Sheet (SDS), which is available upon request.

STORAGE:

This product should be stored in a cool, dry place. It is a powerful oxidizing agent and may cause a fire if it comes in contact with readily oxidizable or combustible matter.

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